



## Preparation, Characterization and Magnetic Property Study of Gd<sup>3+</sup> Substituted Mg-Zn Ferrites

### KEYWORDS

Ferrites, Ceramic method, susceptibility, domain structure, Curie temperature

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**ABSTRACT** Ferrites of  $Zn_xMg_{1-x}Fe_2-yGd_yO_4$  ( $x=0, 0.2, 0.4, 0.8$  and  $1.0$ ;  $y=0, 0.05$  and  $0.1$ ) were prepared by usual solid state standard ceramic method. Characterizations were made by X-ray diffraction method and FTIR spectroscopy. Magnetic properties like temperature dependence normalized AC susceptibility and Curie temperature were studied. This study reveals that  $MgFe_2O_4$  exhibits multi domain structure with high Curie temperature. On substitution of  $Zn^{2+}$ , it is found that Multi Domain (MD) structure changes to Single Domain (SD). Curie temperature also found to decrease due to decrease in A-B interaction. On substitution of  $Gd^{3+}$  in Mg-Zn ferrite system peak obtained in  $MgFe_2O_4$  is suppressed which is attributed to decrease in grain size and further decrease in Curie temperature is attributed to the decrease in B-B interaction. This is because Fe-Fe interaction is greater than Gd-Fe interaction at B-site

### 1. Introduction

In ferrites, magnetic properties like permeability and susceptibility are found to depend on grain size, grain boundary and domain structure [1]. Thus to observe to changes in the internal structure of ferrites this study is more important. Soft ferrites consists of three types of domains like, Multimomain (MD), single domain (SD) and superparamagnetic (SP) particle, which mainly depends on substitutions [2]. These domain states can be distinguished by using the technique of low field AC susceptibility [3]. The MD particles have domain walls [4] and magnetic changes takes place due to domain wall (DW) motion. As particle size decreases, formation of domain walls becomes energetically unfavorable, then it is said to be single domain (SD) particle. In these magnetic changes do not takes place through DW motion but require the rotation of spins resulting in larger coersivity. As the particle size further decreases, spins are affected by thermal fluctuations and the system becomes SP particle. SP particle nature reduces magnetic character of the material.  $Cd^{2+}$  substitution is interesting substitution in the spinels [5]. Addition of  $Cr^{3+}$  in  $NiFe_2O_4$  the domain structure changes from MD to SD [6].  $Al^{3+}$  substituted mixed Cu-Cd ferrites exhibit mixture of SD and MD particles [7].

In this paper we investigated the domain nature in  $Gd^{3+}$  substituted and unsubstituted Mg-Zn ferrite system using low field AC susceptibility measurements.

### 2. Experimental details

Spinel ferrites with general formula  $Zn_xMg_{1-x}Fe_{2-y}Gd_yO_4$  ( $x=0, 0.2, 0.4, 0.6, 0.8$  and  $1.0$ ;  $y=0, 0.05$  and  $0.10$ ) were prepared by standard ceramic method using AR grade oxides of  $Fe_2O_3$ ,  $MgO$ ,  $ZnO$  and  $Gd_2O_3$ . These pure oxides were accurately weighed accordingly to weight ratio required in the final proportions on single pan microbalance. The same were mixed together and wet milled using acetone base. Dried powder of samples was prsintered at  $700^\circ C$  for 10 hours and sintered at  $1050^\circ C$  for 24 hours, cooled and powdered. The pellets of samples were formed by applying  $10^6$  Kg  $cm^{-2}$  using hydrolic press. The pellets were again sintered at  $1050^\circ C$  for 24 hours for better compaction.

The powdered samples were characterised by XRD on philips computerised X-ray diffractometer (PW 3710) using  $CuK$  radiation. The AC susceptibility measurements of polycrystalline ferrite sample was made on Helmholtz double coil set up operated at 260 Hz with constant field of 7 Oe, in the temperature range 300K to 800K. Platinum-Rhodium

thermocouple is used to measure temperature of the powder sample. The Curie temperature of all the pelletized samples was measured by using modified Lorria-Sinha method.

### 3. Results and discussion

#### 3.1 Characterization

Study of X-ray diffraction reveals that all the compositions under investigation were face centered cubic spinel structure. Typical X-ray diffractogram is presented in Fig.1. It is found that lattice constant increases with  $Zn^{2+}$  concentration. Such increase in the lattice constant due to addition of cadmium content was reported in the literature [5]. This increase in lattice constant is attributed to the difference in ionic radii of  $Zn^{2+}$  ion ( $1.03A^\circ$ ) and  $Fe^{3+}$  ion ( $0.67A^\circ$ ). On substitution on  $Gd^{3+}$  ion, the lattice constant found to decrease. This was also attributed to the difference in ionic radii.

Typical IR absorption spectrum is presented in Fig.2. IR absorption spectra of all the compositions show two dominant absorption bands one around  $600\text{ cm}^{-1}$  ( $n_1$ ) and other at  $400\text{ cm}^{-1}$  ( $n_2$ ) which are characteristic of ferrite family. The absorption band observed around  $600\text{ cm}^{-1}$  ( $n_1$ ) is due to the tetrahedral or A-sites and that of around  $400\text{ cm}^{-1}$  ( $n_2$ ) is due to octahedral or B-sites respectively. This confirms the formation of ferrites under investigation.

#### 3.2 Normalized susceptibility

Typical plot of normalized susceptibility ( $\chi/\chi_{RT}$ ) verses temperature is presented in the Fig.2. From this plot it is seen that for magnesium ferrite, the susceptibility slowly increases and reaches peak value with temperature and suddenly drops to zero. The sudden drop of  $\chi/\chi_{RT}$  curve shows the formation of single phase cubic spinel [8]. The increase in susceptibility with peak values suggests there is existence of multidomain (MD) particles in the material [9]. The peak is found to suppressed with substitution of  $Gd^{3+}$  in  $MgFe_2O_4$  and also Curie temperature ( $T_c$ ) decreases with  $Gd^{3+}$  content. For the composition  $x = 0.2$ ;  $y = 0, 0.05$  and  $0.1$ , susceptibility is found to be independent on temperature upto  $T_c$  and after  $T_c$  it suddenly drops to zero. Such nature of curve indicates that the presence of SD particles in the materials [9]. Joshi et al [10] also reported similar behaviour in Mg-Zn ferrite system. The compositions with  $x = 0.4$  and  $x = 0.6$  for  $y = 0, 0.05$  and  $0.1$  shows exponential decrease in susceptibility indicating SD to SP transition. The composition with  $x = 0.8$  and  $x = 1$  with  $y = 0, 0.05$  and  $0.1$  shows paramagnetic behaviour at and above room temperature. Curie temperatures ( $T_c$ ) obtained from susceptibility plots are presented table 1.

The Curie temperature measurement of all the samples was also been carried out by the method suggested by Loria-Sinha [11] and also presented in the table 1. These values are found to be in good agreement with the values obtained from temperature dependence of normalized susceptibility. On substitution of Zn<sup>2+</sup> in MgFe<sub>2</sub>O<sub>4</sub> Curie temperature found to decrease. This is because substituted Zn<sup>2+</sup> ion invariably occupies tetrahedral (A) site, resulting into decrease in A-B interaction [12]. The composition with x= 0.8 and 1.00 shows paramagnetic behaviour at room temperature, their Curie temperature lies below room temperature. Substitution of Gd<sup>3+</sup> ion, Curie temperature of each composition is found to decrease. This is attributed to dilution of B-B interaction[12]. On substitution Gd<sup>3+</sup> ion occupies B-site replacing equivalent Fe<sup>3+</sup> ions and so also decrease in magnetization at B-site.

**4. Conclusions**

From the study it found that MgFe<sub>2</sub>O<sub>4</sub> exhibit multi domain structure and this structure is greatly affected by foreign substitution of Zn<sup>2+</sup> and Gd<sup>3+</sup> ions in it. The study shows that on substitution of Zn<sup>2+</sup>, domain structure changes from MD to SD while on substitution of Gd<sup>3+</sup> it changes from SD to SP. On substitution of Gd<sup>3+</sup>, peak obtained in the graph of normalized susceptibility of MgFe<sub>2</sub>O<sub>4</sub> is suppressed due to the decrease in grain size. Curie temperature a magnetic property also depend on interactions among the ions was found to decrease on substitution of Zn<sup>2+</sup> and Gd<sup>3+</sup>, which is attributed to the dilution of A-B interaction as well as B-B interactions. In conclusion we can say that substitution of non magnetic ions like Zn<sup>2+</sup> and Gd<sup>3+</sup> magnetic properties decreases and also particle size decreases.

**Fig.1 Typical X-ray diffractogram of Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>Gd<sub>y</sub>O<sub>4</sub> ferrites with x=0.4 and y=0.05.**

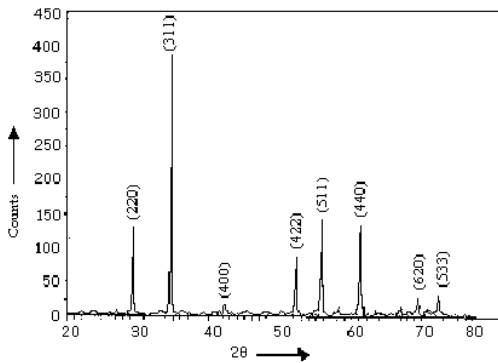


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**Fig.2 Typical IR absorption spectrum of Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>Gd<sub>y</sub>O<sub>4</sub> ferrite system with x = 0, y =0.**

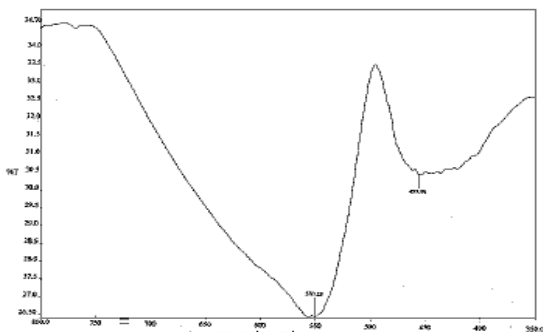


Fig. 2. Typical IR absorption spectrum of Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>Gd<sub>y</sub>O<sub>4</sub> ferrite system with x = 0, y =0.

**Fig.3 Plots of normalized susceptibility (χ/χ<sub>RT</sub>) versus temperature Cd<sub>x</sub>Mg<sub>1-x</sub>Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>Gd<sub>y</sub>O<sub>4</sub> ferrites with y = 0.05.**

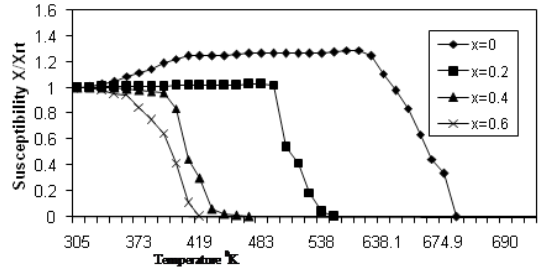


Fig.3 Plots of normalized susceptibility (χ/χ<sub>RT</sub>) versus temperature Cd<sub>x</sub>Mg<sub>1-x</sub>Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>O<sub>4</sub> with y = 0.05.

**Table 1. Curie temperature data from susceptibility measurement and Loria-Sinha method for the composition Zn<sub>x</sub>Mg<sub>1-x</sub>Fe<sub>2-y</sub>Gd<sub>y</sub>O<sub>4</sub>.**

Conc. Zn <sup>2+</sup> (x)	Conc. Gd <sup>3+</sup> (y)	T <sub>C</sub> from AC susceptibility measurement °K	T <sub>C</sub> from Loria-Sinha Method. °K	Frequency of vibration cm <sup>-1</sup>	
				v <sub>1</sub>	v <sub>2</sub>
0.00	0.00	700	711	571	434
		615	602	575	472
		475	490	550	471
		435	455	563	466
		-----	-----	556	451
		-----	-----	540	441
0.00	0.05	659	670	541	434
		540	565	540	431
		455	435	581	470
		410	415	568	465
		-----	-----	559	437
		-----	-----	550	457
0.00	0.10	623	620	623	423
		527	527	604	425
		440	433	590	471
		397	390	586	467
		-----	-----	574	456
		-----	-----	559	439

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