



Podophyllum Hexandrum Fruit Extract as Corrosion Inhibitor of Mild Steel in 1N HCl

KEYWORDS

podophyllum hexandrum fruit extract, Adsorption, mild steel, Inhibition Efficiency

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ABSTRACT

The effect of podophyllum hexandrum fruit extract as an inhibitor for mild steel in 1N HCl using weight loss method has been investigated. The corrosion coupons were immersed in 1N HCl solution containing varying inhibitor concentrations (1%, 2%, 3%, 4% and 5% v/v) within a period of 1-4 hours. From the result, it was found that the adsorption of podophyllum hexandrum fruit extract reduced the corrosion rate in acid medium. The inhibitive action of this plant extract has been found to be significant. The most suitable inhibitor concentration was found to be 5% for an inhibition efficiency of 89%. The inhibition is attributed to physical adsorption. The adsorbed molecules of the inhibitor lie on the surface of the mild steel blocking the active corrosion sites on the alloy. This makes the alloy to have a higher corrosion resistance in the studied environment.

INTRODUCTION

The use of inhibitors to secure metal against corrosion has come under severe criticism. Large amounts of sulphuric acid are being used in the chemical industry for the removal of undesired scales and rust (1). In chemical and allied industries mild steel has been extensively used. One of the ways of protecting mild steel from corrosion is the use of organic inhibitors (2). The use of natural products as corrosion inhibitor is becoming the subject of extensive investigation due to the low cost and eco-friendliness of these products and their biodegradability. This method is fast replacing the synthetic and expensive hazardous organic inhibitors (3). The yield of these compounds as well as their corrosion inhibition abilities varies widely depending on the part of the plant and its location (4). The extracts from the leaves, seeds, heart wood, bark, root and fruit of plant have been reported to inhibit metallic corrosion in acid media (5). The corrosion inhibitive action of flower extracts of Magnolia champaca flower on mild steel corrosion in 0.5 M H₂SO₄ solution was studied (6). The inhibition efficiency of sodium molybdate (SM) - Zn²⁺ system in controlling corrosion of carbon steel in seawater has been evaluated by weight-loss method(7). The inhibition behavior of Momordica charantia seeds extract (MCSE) as an environmentally benign corrosion inhibitor for J55 steel was investigated in 3.5 wt.% NaCl saturated with CO₂ solution(8). The corrosion and inhibition behaviors of mild steel in sulfuric acid + 5% EtOH in the presence of tannin extract of Chamaerops humilis plant (LF-Ch) and potassium iodide (KI) have been studied(9). The corrosion inhibition of mild steel in hydrochloric solution by Black pepper extract (Piper nigrum fem. Piperaceae) was studied(10). The inhibitive effect of the aqueous extract of Coffee senna (CS) on the corrosion of mild steel in 1M HCl and 0.5M H₂SO₄ solutions was investigated(11). An improved inhibition efficiency of 95.36% was observed with the mixture of Nypa fruticans' wurmb extract and KI at 30°C in 0.5M HCl(12). The inhibition of the corrosion of mild steel in aqueous solutions by ethyl acetate extract of Uncaria gambir containing catechin has been studied. Inhibition was found to be highest at a concentration of 150 ppm in solutions with a pH of 5(13). Acetone extract of red peanut skin

gave an excellent protection on mild steel in 2M hydrochloric acid at 303K – 323K(14). The corrosion inhibitive action of flower extracts of Magnolia champaca flower on mild steel corrosion in 0.5 M H₂SO₄ solution was studied. The results obtained indicate that the extracts functioned as good inhibitors in 0.5 M H₂SO₄ solution(15). Silver nanoparticle using Cleistanthus Collinus Leaf Extract has been reported(16-17). The aim of the present work is to study the effect of the fruit extract of Podophyllum Hexandrum on the corrosion rate of mild steel in HCl acid using weight loss measurements and Electrochemical impedance studies.

EXPERIMENTAL

Materials

Methanol (Aldrich chemicals), Distilled water, Podophyllum hexandrum fruit extract and HCl (Qualigens chemicals).

Preparation of specimens

The Mild steel specimens with a dimension of 4×2×0.2 cm were taken. The specimens were mechanically polished by using different grades of SiC(grades 120-400-600) emery papers, degreased with acetone, dried and then stored in desiccator before use.

Preparation of extract

Fresh fruits of the podophyllum hexandrum were washed under running water. Exactly 300g of the fruit was taken. This was crushed and extracted using 600 ml methanol for a duration of 72 hrs at 60°C to obtain 50% of the solution. This methanolic extract was used for the corrosion studies.

Weight loss method

The Mild Steel was weighed and tied with threads and suspended in 100 ml HCl solution, with and without the extract of different concentrations at different intervals. Duplicate experiments were performed in each case and the mean value of the weight loss has been recorded. The weight loss has been calculated using the formula:

Corrosion rate = $87.6 \times W / \text{Surface area of the specimens}$

x T x D

Where, W = Weight loss in mg, A = Area of specimens in SqCm, D = Density of specimens in gm/Cm³ and T = Time for which the specimens were exposed to the corroding medium (in hours)

Results and Discussion

The Corrosion behavior of MS substrate in 1N HCl in the absence and in the presence of extract was studied at different concentrations of extract and different time intervals. The variation of the corrosion rates with time of exposure for the reference and five different inhibitor concentrations in 1N HCl solution (Tables 1-4) has been studied. The inhibition

efficiency varies with time of exposure for different inhibitor concentrations in 1N HCl solution.

The significant decrease in corrosion rate at 5% inhibitor concentration can be attributed to the adsorption of molecule of the inhibitor on the alloy surface, since this inhibitor contains tannin which acts as physical barrier to restrict the diffusion of ions to and from the alloy and then prevent the alloy atoms (ions) from participating in further anodic or cathodic reactions, thus resulting in decrease in the corrosion rate [18]. The plant extract can adsorb on the alloy surface and block the active sites on the surface, thereby reducing the corrosion rate in the medium.

Table 1 Corrosion behavior of mild steel in 1N HCl in the absence and presence of different concentrations of podopyllum hexandrum fruit extract. Immersion time 1 hr.

| S.No | Concentration Of inhibitor(ml) | Initial weight of the substrate (g) | Final weight of the substrate (g) | Difference in weight (g) | Rate of Corrosion (mpy) | I.E(%) |
|------|--------------------------------|-------------------------------------|-----------------------------------|--------------------------|-------------------------|--------|
| 1 | Blank | 5.717 | 5.713 | 0.004 | 10.3070 | - |
| 2 | 1% | 5.623 | 5.620 | 0.003 | 7.7302 | 25.0% |
| 3 | 2% | 5.788 | 5.784 | 0.003 | 7.6085 | 25.0% |
| 4 | 3% | 4.846 | 4.844 | 0.002 | 5.1535 | 50% |
| 5 | 4% | 5.283 | 5.281 | 0.002 | 5.15 | 50% |
| 6 | 5% | 4.905 | 4.904 | 0.001 | 2.5767 | 75% |

Table 2 Corrosion behavior of mild steel in 1N HCl in the absence and presence of different concentrations of Podopyllum Hexandrum fruit extract. Immersion time 2 hrs.

| S.No | Concentration Of inhibitor (ml) | Initial weight of the substrate(g) | Final weight of the substrate (g) | Difference in weight(g) | Rate of corrosion (mpy) | I.E(%) |
|------|---------------------------------|------------------------------------|-----------------------------------|-------------------------|-------------------------|--------|
| 1 | Blank | 5.893 | 5.888 | 0.005 | 6.443 | - |
| 2 | 1% | 5.309 | 5.305 | 0.004 | 5.153 | 20.0% |
| 3 | 2% | 4.916 | 4.913 | 0.003 | 3.865 | 40.0% |
| 4 | 3% | 5.295 | 5.292 | 0.003 | 3.865 | 40.0% |
| 5 | 4% | 5.106 | 5.086 | 0.002 | 2.576 | 60% |
| 6 | 5% | 5.562 | 5.561 | 0.001 | 1.288 | 80% |

Table 3 Corrosion behavior of mild steel in 1N HCl in the absence and presence of different concentrations of podopyllum hexandrum fruit extract. Immersion time 3 hrs.

| S.No | Concentration Of inhibitor(ml) | Initial weight of the substrate (g) | Final weight of the substrate (g) | Difference in weight(g) | Rate of corrosion (mpy) | I.E(%) |
|------|--------------------------------|-------------------------------------|-----------------------------------|-------------------------|-------------------------|--------|
| 1 | Blank | 4.807 | 4.800 | 0.007 | 6.012 | - |
| 2 | 1% | 4.93 | 4.925 | 0.005 | 8.865 | 28% |
| 3 | 2% | 5.558 | 5.554 | 0.004 | 3.220 | 42.8% |
| 4 | 3% | 4.950 | 4.947 | 0.003 | 2.576 | 57.1% |
| 5 | 4% | 4.902 | 4.9 | 0.002 | 1.932 | 71.4% |
| 6 | 5% | 4.838 | 4.837 | 0.001 | 0.644 | 85.7% |

Table 4 Corrosion behavior of mild steel in 1N HCl in the absence and presence of different concentrations of podopyllum hexandrum fruit extract. Immersion time 4 hrs.

| S.No | Concentration Of inhibitor(ml) | Initial weight of the substrate (g) | Final weight of the substrate (g) | Difference in weight(g) | Rate of corrosion (mpy) | I.E(%) |
|------|--------------------------------|-------------------------------------|-----------------------------------|-------------------------|-------------------------|--------|
| 1 | Blank | 4.807 | 4.800 | 0.007 | 6.012 | - |
| 2 | 1% | 5.531 | 5.525 | 0.006 | 3.865 | 35.7% |
| 3 | 2% | 5.868 | 5.863 | 0.005 | 3.220 | 46.4% |
| 4 | 3% | 4.432 | 4.428 | 0.004 | 2.576 | 57.1% |
| 5 | 4% | 4.203 | 5.200 | 0.003 | 1.932 | 67.8% |
| 6 | 5% | 4.623 | 4.622 | 0.001 | 0.644 | 89.2% |

Conclusions

The plant extract (podopyllum hexandrum fruit extract) decreases the corrosion of mild steel in 1N HCl solution to an extent of 89% at the level of 5% inhibitor concentration upto four hours. The interaction between the inhibitor and the alloy results in physical adsorption. The adsorbed inhibitor molecules are attached at the alloy surface thereby blocking active corrosion sites hence lowering corrosion rate.

REFERENCE

1. Prabhu R.A., Venkatesha T. V., Shanbhag A. V., Kulkarni G.M, and Kalkhambkar R.G, Inhibition effects of some Schiff's bases on the corrosion of mild steel in HCl Solution, *Corrosion Science*, 2008, 50, 3356 – 3362. | 2. Hosary A.A.E., Saleh R.M., 1-Dahan H.A.E (1990): Corrosion and Scale inhibition in cooling water by naturally Occurring Substance, 7th edition *On Corrosion Inhibitors* (Ferrara,Italy: University of Ferrara), p 725. | 3. Arab S.T. and Noor E.A, Inhibition of Acid Corrosion of Steel by Some S-Alkylisothiuronium Iodides, *Corrosion science*, 1993, 49, 122. | 4. Okafor P. C., Ikpi M.O., Uwah I.E., Ebenso E.E., Ekpe U.J and Umoren S.A, Inhibitory action of Phyllanthus amarus extracts on the corrosion of mild steel in acidic media, *Corrosion Science*, 2008, 50, 2310. | 5. Onuegbu T. U., Umoh E.T., Onuigbo U. A. Eupatorium Odoratus As Eco-Friendly Green Corrosion Inhibitor Of Mild Steel In Sulphuric Acid, *International Journal Of Scientific and Technology Research*, 2013, 2, 4. | 6. Iloamaeke I. M, Onuegbu T. U., Umeobika U. C and Umedum N. L. "Green Approach to Corrosion Inhibition of Mild Steel Using Emilia Sonchifolia and Vitex Doniana In 2.5M HCl Medium" *International Journal of Science and Modern Engineering*, 2013, 1, 10. | 7. S. Rajendran, K. Anuradha, K. Kavipriya, A. Krishnaveni, and J. Angelin Thangakani "Inhibition Of Corrosion Of Carbon Steel In Sea Water By Sodium Molybdate – Zn²⁺ System", *Eur. Chem. Bull.*, 2012, 1, 503. | 8. Ambrish Singh, Yuanhua Lin, Wanying Liu, Deng Kuwanhai, Eno. E. Ebenso, Jie Pan, "Application of a Natural Inhibitor for Corrosion Inhibition of J55 Steel in CO₂ Saturated 3.5% NaCl Solution" *Int. J. Electrochem. Sci.*, 2013, 8, 12851. | 9. O. Benali, H. Benmehdi, O. Hasnaoui, C. Selle, R. Salghi, "Green corrosion inhibitor: inhibitive action of tannin extract of Chamaerops humilis plant for the corrosion of mild steel in 0.5M H₂SO₄" *J. Mater. Environ. Sci.*, 2013, 4, 127. | 10. M.A. Quraishi*, Dileep Kumar Yadav and Ishtiaque Ahamad, "Green Approach to Corrosion Inhibition by Black Pepper Extract in Hydrochloric Acid Solution" *The Open Corrosion Journal*, 2009, 2, 56. | 11. Chris O Akalezi, Conrad K Enenebaku and Emeka E Oguzie "Application of aqueous extracts of coffee senna for control of mild steel corrosion in acidic environments" *International Journal of Industrial Chemistry*, 2012, 3, 13. | 12. Orubite-Okorosaye, K; 1jack, I R; 1ochei, M; Akaranta, O," Synergistic Effect of Potassium Iodide on Corrosion Inhibition of Mild Steel in HCl Medium by Extracts of Nypa Fruticans' Wurmb", *J. Appl. Sci. Environ. Manage.*, 2007, 11, 27. | 13. Mohd. Hazwan Hussin and Mohd. Jain Kassim, "Electrochemical Studies of Mild Steel Corrosion Inhibition in Aqueous Solution by Uncaria gambir Extract" *Journal of Physical Science*, 2010, 21, 1. | 14. S.Ananth Kumar, A.Sankar, M.Vijayan, S.Rameshkumar, "Magnolia Champaca- Flower Extracts As Corrosion Inhibitor For Mild Steel In Acid Medium" *IOSR Journal of Engineering* 2013, 3,10. | 15. Iloamaeke I. M, Onuegbu T. U., Umeobika U. C., Umedum N. L., "Green Approach to Corrosion Inhibition of Mild Steel Using Emilia Sonchifolia and Vitex Doniana In 2.5M HCl Medium" *International Journal of Science and Modern Engineering* 2013. 1, 12. | 16. R.Vennila, P.Kamaraj, M.Arthanareeswari,B.Sivakumar, Green Synthesis of Silver Nanoparticles from Cleistanthus Collinus Leaf Extract and Their Biological Effects *International Journal of Chemistry*, 2013, 34, 1103. | 17. R. Vennila, P Kamaraj and Arthanareeswari M, Corrosion Behavior of Ag Doped Bi₂O₃ Nano Composite, *Chem Sci Trans.*, 2013, 2(S1), S173. | 18. C. A. Loto, "The Effect of Bitter leaf on the Inhibition of Mild Steel in HCl and H₂SO₄," *Corrosion Prevention and Control Journal*, 2003, 50, 43. |